#### Impregnation of Metal Complex into Epoxy Insulation Materials Using Supercritical Carbon Dioxide and Its Application for Copper Plating

Hidebumi Ohnuki – Oriental Printed Circuits Ltd., Hong Kong Shinji Sumi - Guangzhou Meadville Electronics Co., Ltd., China

#### Abstract

Metal plating of epoxy polymer has been widely applied for industrial products for a long time, especially in the field of Printed Circuit Boards (PCB's). This technique is one of the most important technologies of electronics devices with high reliability and guaranteed quality. The authors are developing a new concept of PCB's for next decade generations to improved the techniques of which including more fine line circuitries, higher densities and narrower spaced conductor lines. An essential part of subject for this technology is to improve the weak copper adhesion peel strength on epoxy insulation materials. To obtain the good adhesion property of copper plating, it is now widely used the Pd Colloid Solution method. In order to further improve more excellent adhesion for next decade generation PCB's, we are investigating Super Critical Fluid (SCF) method. In this paper, an attempt has been made to impregnate some metal complexes into epoxy resin and then decomposed the complexes to produce free metal in the resin by reduction . Using the deposited metal is efficient electro-less Cu plating can be achieved. We will discuss the selection of the metal complexes and impregnation conditions on the complexes as well as peel strength of the plating.

#### Introduction

PCB's consumers market is rapidly shift to the high density and high reliability circuit technologies [1] [2]. Especially in Portable Handy (Cellular) Phone and Laptop Computer markets, they're very strongly requesting those latest technologies applications. Those technical demands background, there are challenging to satisfy with smallest size, lightly weight and high multiple functional equipments. Then PCB's apply to change into more fine lines and high density circuits boards. Those fine circuitry condition demands for next decade period PCB's Era will be strongly requesting to excellent copper conductor peel strength and reliability. On those markets requesting background to satisfy with the excellent PCB's quality level, we are investigating on Super Critical Fluid (SCF) method and an attempt has been done to impregnate some metal complexes into epoxy resin and decomposed them to produce free metal in the resin by the reduction of complex.

Currently PCB's main plating technology is the Pd Colloid Solutions process combining with electro-less plating and electric plating for connecting the both side circuits and inner layer circuitries. This plating technology has strong point for cost performance and very popular using, but weak point is not excellent copper peel strength on epoxy direct surface plating.

Using the SCF deposited metal an efficient electro-less Cu plating is achieved [3] [4] [5].

In this paper we are presenting the SCF application development to epoxy insulation which is the first approach to impregnate metal complex. After some pre-experiments a few Pd complexes were selected and suitable conditions for their impregnations into epoxy polymer plate were investigated, considering the peel strength of the Cu plated polymer plate.

#### Experimental

The impregnated material was an Epoxy C stage plate. This epoxy resin for testing was applied with Bis Phenol A-type resin (Liquid type property) which is widely used in the PCB's industries.

Figure 1 shows the Bis Phenol A-type chemical formulation.

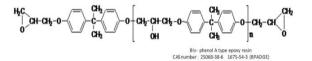


Figure 1 Bis Phenol A-type Epoxy resin (Liquid type)

As impregnation medium, Supercritical Carbon Dioxide [SC (CO<sub>2</sub>)] was used. Temperature, pressure and treatment time were controlled for metal complex impregnation.

Table 1 shows the SC (CO<sub>2</sub>) variable factors combination and conditions for epoxy impregnation test.

Table 1 Impregnation conditions					
	Level1	level2	level3	Level4	
Temperature (Deg.)	60	100	140	150	
Pressure (Mpa)	10	15	20	25	
Time(Min)	30	60	90	120	
Leakage time(Min)	1	10	30		

Table 1	Impregnation	conditions
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Selected Pd complexes were Pd(II) hexafluoroacetylacetonate (Complex A), Bis-acetylacetonate Pd(II) (Complex B) and Pd(II)acetate (Complex C).

Figure 2 shows appearance of the selected complexes before and after  $SC(CO_2)$  treatment and their chemical structures. The epoxy plates were treated under various SC(CO<sub>2</sub>) conditions to impregnate with Pd complexes. Pd complex impregnated epoxy samples were further treated to decompose the complexes to produce free metal Pd. Then Pd impregnated epoxy plates were applied through electro-less and electric copper sulfate plating processes for metallization. Impregnated Pd complex samples and plated samples are checked by PST (Peel Strength Test), SEM (Scanning Electron Microscope), XPS (X-ray Photoelectron Spectroscopy), EDX (Energy Dispersive X-ray Fluorescence Spectrometer) and TEM (Transmission Electron Microscope) to obtain the Pd complex conditions in the epoxy material and to compare the currently plating peel strength.

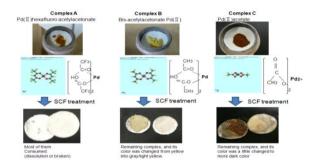


Figure 2 Three typical type of complexes conditions before and after SC (CO<sub>2</sub>) treatments

#### **Results and Discussion**

We will discuss in this paper, if the SC (CO<sub>2</sub>) is effective or not in existence for Pd complexes impregnation into epoxy materials and which kind of complexes is effective, for high amount of complex impregnation into epoxy material to obtain high performance Cu plating.

As the first step, we checked the effect of SC (CO<sub>2</sub>) to epoxy materials and to the adsorption of complex on epoxy material.

Firstly, a blank treatment of epoxy material without both SC (CO<sub>2</sub>) and complexes on epoxy material was checked. Secondly, epoxy material was treated with SC  $(CO_2)$  and without complexes. Thirdly, the materials were treated with both SC  $(CO_2)$  and some complex.

Figure 3 shows one of the test results.

Excellent results were found with good copper metal plating performances using SC (CO<sub>2</sub>). In the other two processes, poor the copper metal plating on the epoxy resin surfaces was given.

It is clear that SC (CO<sub>2</sub>) is one of good technical approach for impregnating the complexes onto epoxy materials, and also  $SC(CO_2)$  makes better copper plating adhesion performances compared with blank treatments.

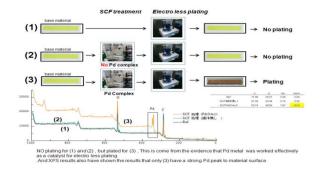
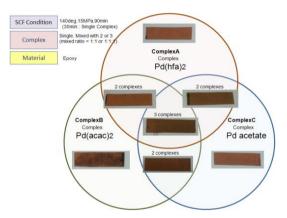


Figure 3 Verification check for impregnation test.

As the second step, the complex impregnation using  $SC(CO_2)$  was carried out for three kinds of complexes for longer time. We used complexes singly or mixed parameter. The Pd complex impregnated samples were put into the electro-less plating solution.

Figure 4 shows the appearances of electro-less copper plating after impregnation of one kind of complexes or some mixture complexes.



#### Figure 4 Electro-less plating visible conditions after three complexes combination impregnations

Cu plating was carried out easily. To increase the thickness of the Cu plating layer, the electric plating has been done by treating the electro-less plated plate into copper sulfate plating solution with 2 ASD for 30 minutes. The thickness of the Cu plating was controlled around  $15\mu m$ .

**Figure 5** shows the results of copper peel strength test after electric copper plating. In the case of single complex Pd (II) acetate (complex C), the highest peel strength of copper plate was obtained. In the case of mixed state with two or three mixed complexes, good peel strength was found by mixture of Pd (II) hexafluoroacetylacetonate and Pd (II) acetate (Complex A+C).

It is very interesting that better copper peel strength was obtained in the case of mixed state with 2 complexes comparing in the case with single complex impregnation.

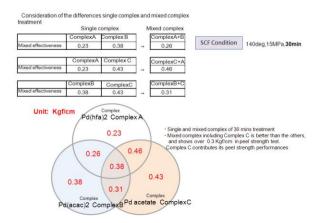


Figure 5 Copper plating average peel strength deviations for three complexes impregnations territory

Finally, we compared Cu plated sample using  $SC(CO_2)$  and the current plating method. As results, the better copper peel strength by Cu plating applying  $SC(CO_2)$  method was obtained comparing with the current plating method (Pd Colloid solution plating).

The stronger peel strength complex was illustrated mainly in the case of Pd (II)acetate (Complex C) or a mixture of Pd(II)hexafluoroacetylacetonate and Pd(II)acetate (Complex A+C). From XPS analysis, we found the metal state of Pd was not located on the polymer surface, but in a little deeper region of the polymer plate.

**Figure 6** shows XPS analysis for mixed complexes (A+C) which applied impregnated condition and after Ar gas etching treatments removing skin surface contaminations. Pd intensity strength is a little bit increased as results. Then Pd complexes are located on near the skin surfaces on Epoxy materials.

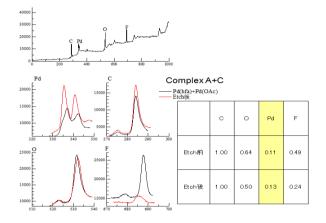


Figure 6 XPS analysis of mixed complex (A+C)

Figure 7 shows EDX mapping conditions result for mixed complex (A+C). Pd is also presented epoxy surface and very strongly and naturally existing on resin surfaces.

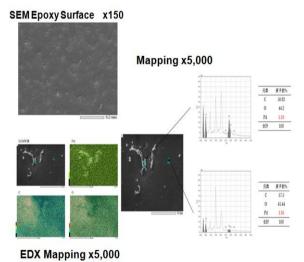


Figure 7 Pd EDX mapping condition of Complex (A+C)

Finally we observed TEM analysis for visible check.

Figure 8 shows TEM cross section analysis. The Epoxy surface Pd exists on both shiny surface and mat surface.

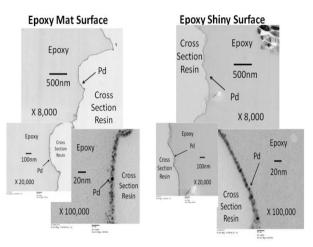


Figure 8 Pd Impregnation condition on Epoxy by TEM

Pd thickness is 20nm deep on the skin surfaces. This 20nm impregnation thickness was called *Skin Impregnation*. We got the Pd complex impregnation area on Epoxy skin surfaces.

In **Figure 9**, copper plating peel strength was summarized for each Pd complex impregnation and current plating cases. This peel strength level is approximately 0.4~0.5kgf/cm. These values are about 2.0~2.5 times stronger, compared than current sulfate copper direct plating peel strength on bare epoxy surfaces plating.

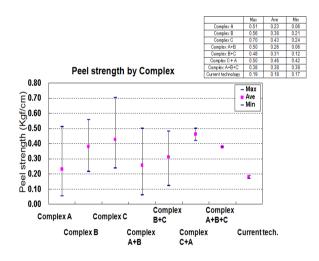


Figure 9 SC (CO<sub>2</sub>) and current Cu plating peel strength.

Figure 10 shows all of the summarized results. Better Cu plating peel strength was achieved by using SC  $(CO_2)$  with both selected SCF conditions and selected metal complexes.

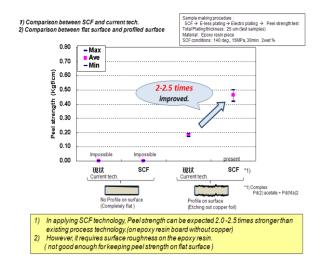


Figure 10 SC(CO<sub>2</sub>) and current Cu plating peel strength and it surface conditions impact

But there are some un-expected results that are impacted Epoxy surface treatment conditions. It is clear that profile surface (mat) is good for adhesion result and non profile surface (shiny) is poor or not peel strength result by  $SC(CO_2)$  impregnations. Pd complex can be impregnated onto both surfaces by  $SC(CO_2)$ . But the copper peel strength is affected from Epoxy surface materials roughness. Plated copper peel strength is affected from its anchor phenomenon. These anchor phenomena consisted by Epoxy surface roughness.

#### Conclusions

We found some of good metal complexes for impregnation into epoxy insulation plate using  $SC(CO_2)$  treatment. The selected metal complexes are Pd(II)acetate (Complex C) for single complex usage, and for mixed use case, Pd(II) hexafluoroacetylacetonate and Pd(II)acetate (Complex A+C).

As the best case, we choose the mixed complexes combination. It is considered that the dissolution of the complexes in  $CO_2$  and the affinity of the complexes are more important for enough impregnation into epoxy resin.

The best impregnation condition was 140°C, 15 MPa and 30 min. At lower temperature and lower pressure is not enough Cu plating. Under those conditions, not satisfied Cu plating was obtained.

The better peel strength of the Cu plating can be obtained using  $SC(CO_2)$  method, not conventional Pd Colloid Solution processes. Currently this  $SC(CO_2)$  process has some undefined technical items, such as not clearly fixed to the surface on epoxy surfaces.

Pre-surface treatment levels before applying the  $SC(CO_2)$  cannot be and the volume of the complexes amount into epoxy impregnation cannot measured

#### References

- H.Ohnuki and S.Sumi, "JPCA 2008 Seminar", the 2nd advance electronics pavilion seminar, P71~P88, June 11, 2008.
- [2] H.Ohnuki and S.Sumi, "HKPCA & IPC 2008" International Technical Conference, Dec. 4, 2008.
- [3] T.Hori and et al, "Introduction for Super Critical

Fluid", Maruzen Tokyo, p.176~ (Dec. 31, 2008).

 [4] H.Ohnuki and et al, "Impregnated the Compound complex Initiators into Epoxy Insulation Materials by SC(CO<sub>2</sub>) methods and analysis for the Copper Plating Peel Strengths Evaluations" June Annual meeting of The Society of Fiber Science and Technology, Tokyo, Vol.64,No.1 p.262 (June 11, 2009).

[5] Japan Patent, "JP 2006-131769 A 2006.5.25".

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#### Acknowledgements

The work based on research done by Meadville Enterprises (HK) Limited and Graduate School of Engineering, University of Fukui under agreement with Mutual Research and Development Agreements (MRDA).



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# Using Supercritical Carbon Dioxide & Its Application for Copper Plating

Meadville Enterprises (HK) Limited

美維企業(香港)有限公司

a member of TTM Technologies, Inc.

Present by: Hidebumi Ohnuki

Co-present by: Shinji Sumi

2011-04-13~15

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## Many thanks to USA's great *disaster relief* assistant. *Tohoku-Pacific Ocean Earthquake/Tsunami*



Tsunami strike northern Japan coast area





Car on the roof after tsunami disaster



US president Mr. Obama make condolences



Great assistant by US army, and navy



# Japan

# Quake / Tsunami / Nuclear crisis

Japan faces unprecedented crisis (11 Mar. 2011)

- ✓ Great earthquake (Magnitude 9.0)
- ✓Tsunami
- ✓Nuclear plant disaster
  - --Scheduled blackout to save power
  - --Radioactive food, water

We never give up and launch reconstruction soon !! I would like to express deepest appreciation to US citizen, government, military. Also about 134 countries and 39 international organizations. We hope you to support

the continuous cordial assistants.



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# **Application industry of PCB**



#### Important role as a basis of various industries

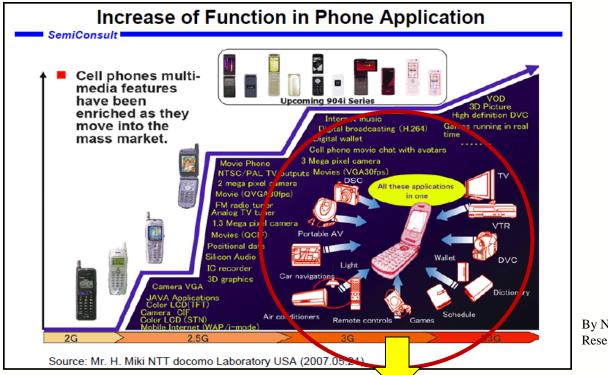


## **PCB application example**

	Past	Present	Future
Cellar Phone	Analog	High speed / Data transformation 3G	Multi-function/ Huge volume Data communication 3.5G,4G
Car	Electronic ratio≒0%	Electronic ratio≒40%	Eco car Electronic ratio>80%



## **Background : Technology trend & Customer requirement**



By NTT docomo Researcher.

In mobile phone development and evolution, following items are required. ✓ Most advanced technologies with high density and downsizing level. ✓ And not only in HDI but also in Rigid-Flex, module package. ✓ High quality and reliability with mass pro. level. ✓ Tough cost reduction against competitors

### **Research Mobile phone trend** → **Strong hint for R&D activities**



## **Technology trend & Customer requirement**

Portable, Lighter, Any layer HDI,

**Required property** 

## Key word

Huge data transaction (Video streaming), Low transmission loss, embedded tech, Optical tech, high frequency

Downsizing

**High Speed** 

IC downsizing

**High reliability** 

**Inspection & repair** 

Registration

**Cost reduction** 

**Green electronics** 

High density design( Fine line, Small via, Fine pitch CSP,LDD)

Chip size downsizing

High reliability material & process, Simulation analysis, New concept assembly tech,

High accurate inspection (4 wire test, AOI,,), repairing tech (Op,Sh for fine design)

Stable dimension material, LDI, LDD, Scale factor data linkage system and auto analysis and feedback.

Simplify process, Improve yield (systematic failure feed back system or machine development)

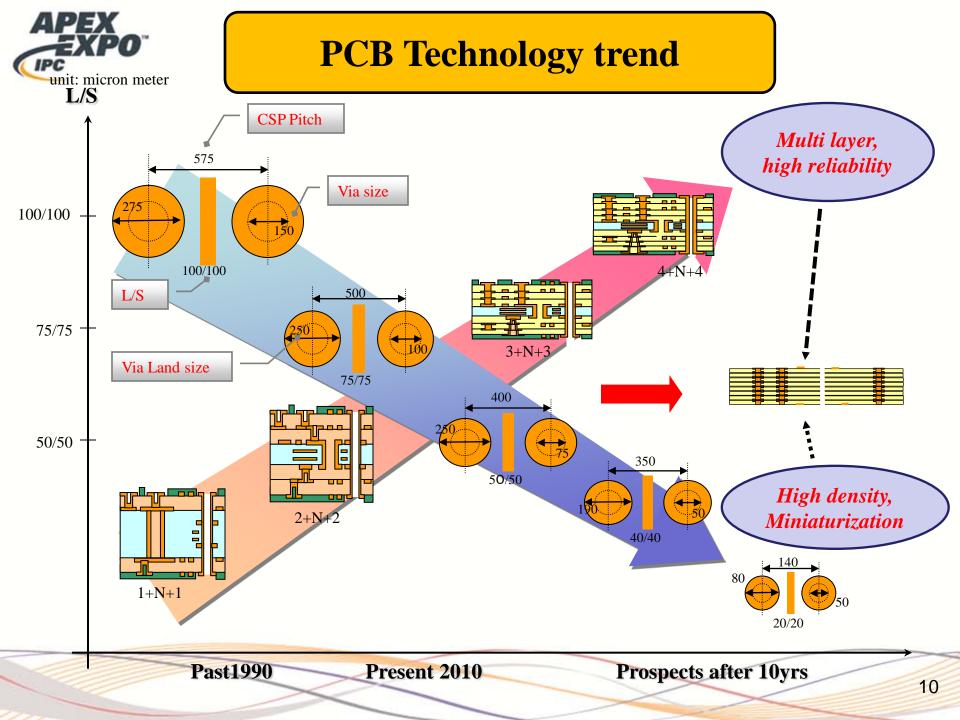
HF, Lead free, World level common target, Brand image



## **NTT docomo Mobile Phone Transmission Speed**

## NTT docomoの携帯電話の通信速度







# Why to use the SCF for PCB?

## Expect to get the

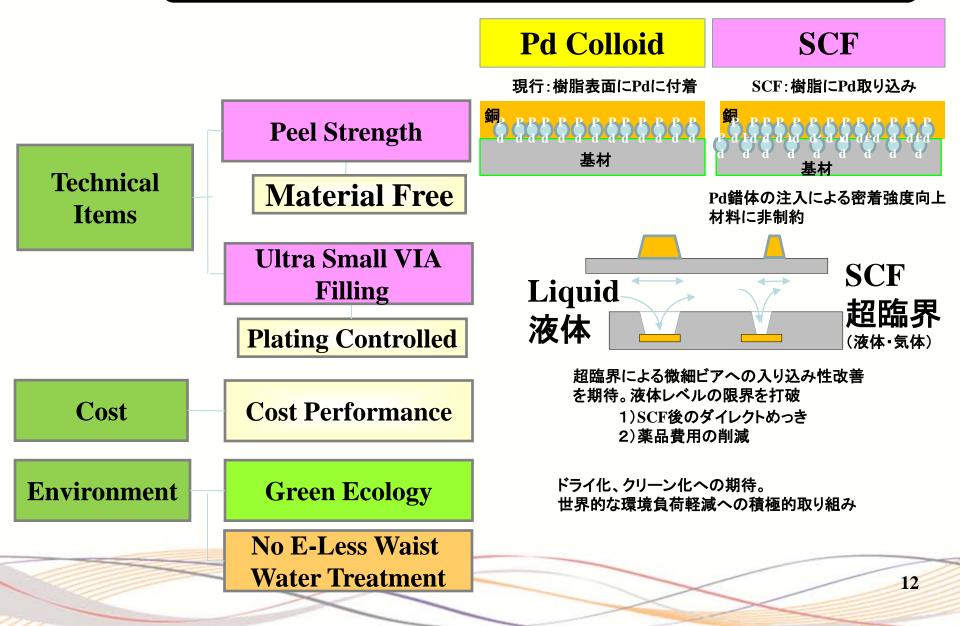
# Good copper peel strength for fine line

# And

Perfect small VIA hole Cu plating









# **SCF:** Super Critical fluid Technology

# **Test details**

- (1) Application SCF technology to Epoxy resin
- (2) SCF treatment condition and Complex selection
- (3) SEM observation on Epoxy surface after SCF
- (4) Measurement of complex impregnation by XPS analysis
- (5) Epoxy EDX Mapping check
- (6) TEM Analysis for Epoxy Matte & Shiny Surface
- (7) Verification peel strength performance



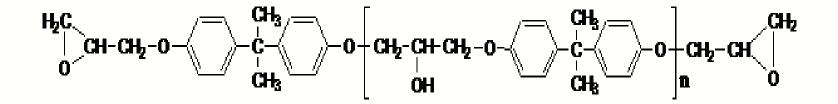
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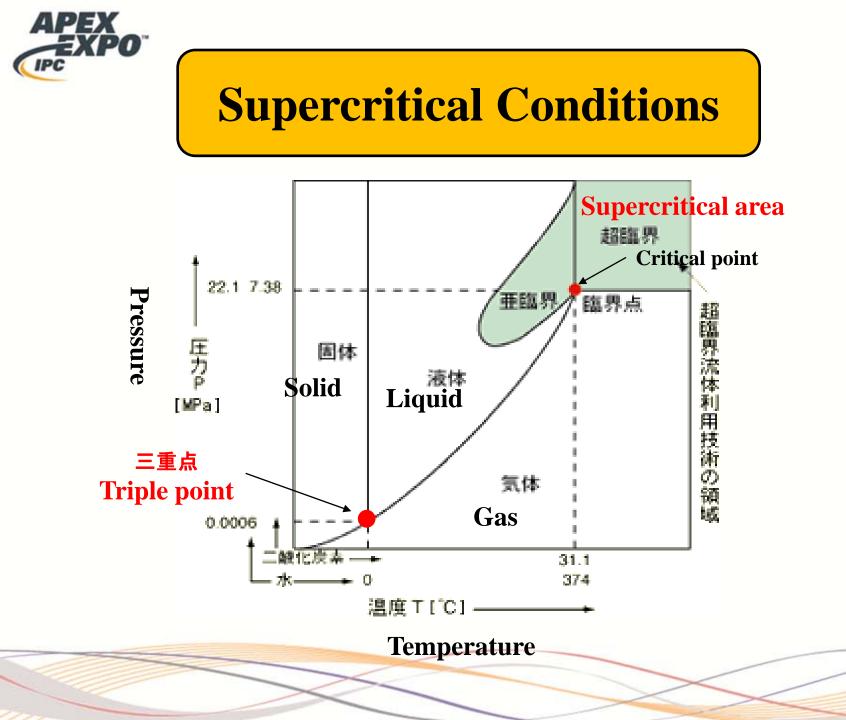
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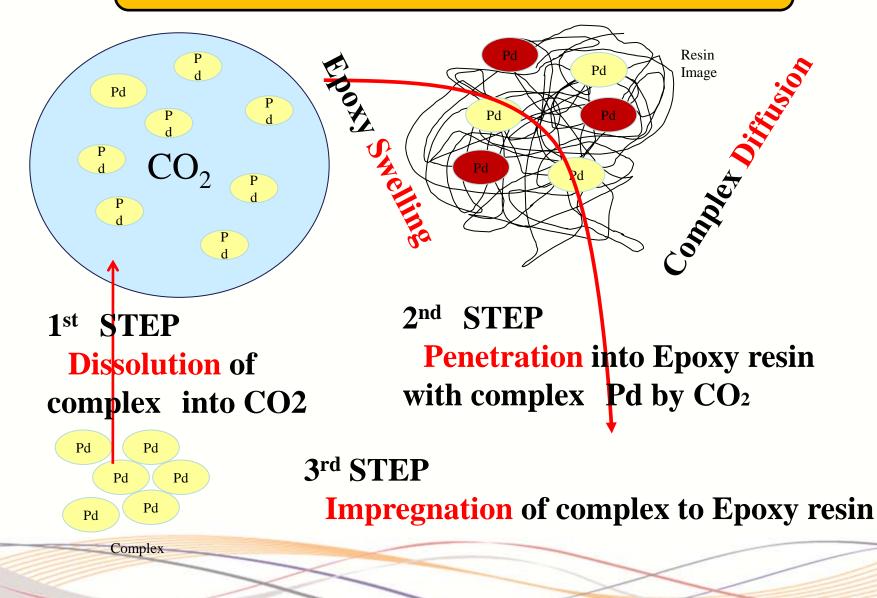
## **Epoxy Material molecular formula and structure**



Bis- phenol A type epoxy resin CAS number: 25068-38-6 1675-54-3 (BPADGE)



# **Mechanism by SCF technology**

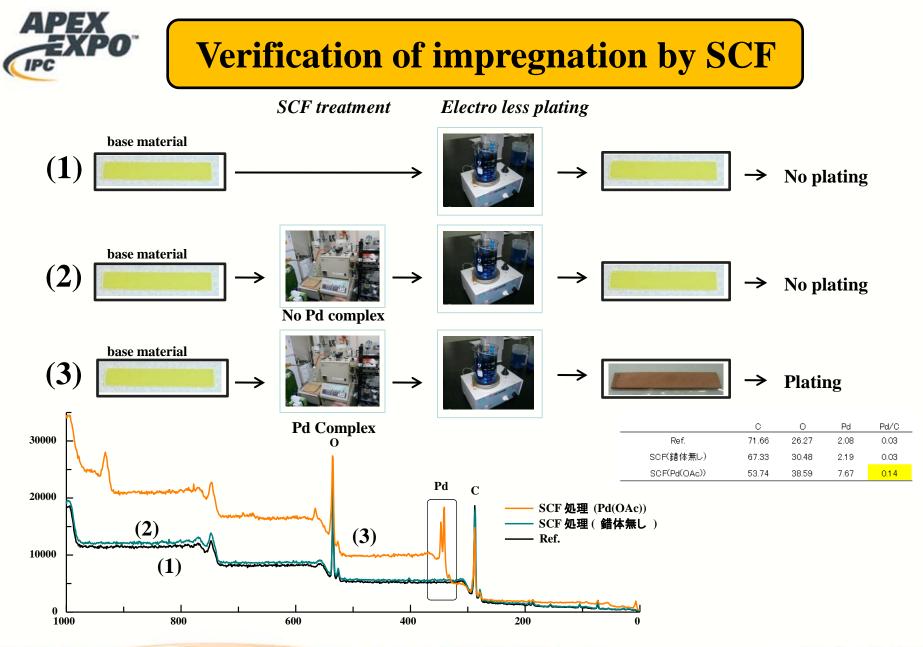


	What is SCF technology ?						
		SCF Super Critical Fluid	Chemical (Current method)				
	Process category	Dry process	Wet process				
		Supercritical 超臨界	Liquid 液体				
Reaction process	High pressure (15MPa)& High temperature(140deg.) In chamber	Temperature (<100deg.) In dipping tank					
		Impregnation	Adhesion				
	Catalyst	Complex powder with metal 金属錯体粉末	Soluble liquid chemical 溶解液体薬品				
	Machine	Special Reacting machine	Chemical treatment tank				

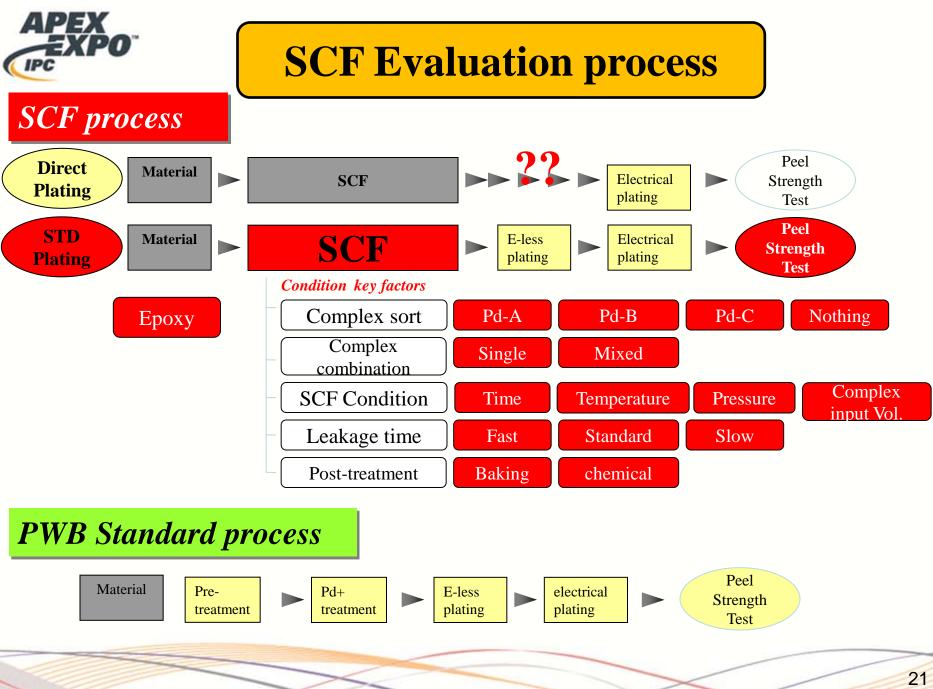


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NO plating for (1) and (2), but plated for (3), This is come from the evidence that Pd metal was worked effectively as a catalyst for electro less plating. And XPS results also have shown the results that only (3) have a strong Pd peak to material surface.





## **SCF Treatment condition**

	Condition1	Condition2	Condition3	Condition4
Temperature ( Deg.)	60	100	140	150
Pressure (MPa)	10	15	20	25
Time(Min)	30	60	90	120
Leakage time(Min)	1	10	30	



# **Test instruments**

## 1) Impregnation treatment by SCF machine



SCF machine : ISCO SFX2-10 Test Condition : Temperature : 140 deg. Pressure : 15MPa ( to 25MPa) Treatment time : 30 min (to 90min)

## 3)Electro plating

## 2)Electroless plating



PTH Chemical : Atotech Temp. 30-35 deg. Time : 5 min Agitation : ON (stirrer)



Plating Chemical : Atotech Temp. 25 deg. (Room Temp.) Time : 50 min (20um Target) Bubbling : ON





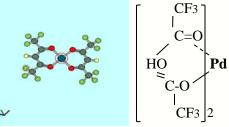


Complex A Pd( II )hexafluoro acetylacetonate



Fiscope-to-to:全路總督統大百方A、東洋松河部博士+Rends Memostar De 12033 0 15-3-14-17 Longi 5 004 Arge565 Divale5 8 Pd

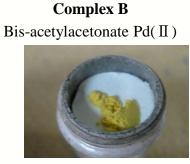
IPC







Most of them Consumed (dissolution or broken)





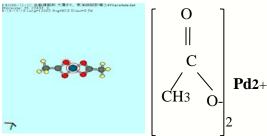




Remaining complex, and its color was changed from yellow into gray/light yellow.

**Complex C** 







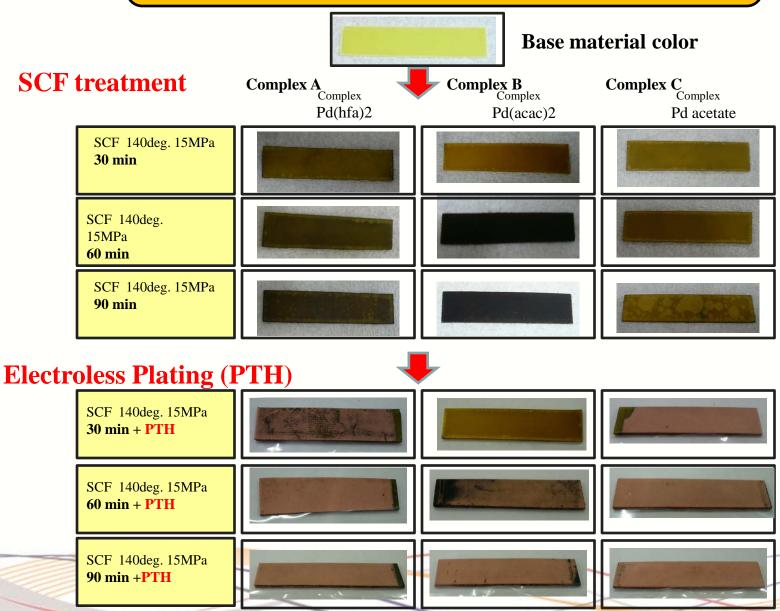


Remaining complex, and its color was a little changed to more dark color

## Color change on material after SCF treatment & Electroless plating

APEX

IPC

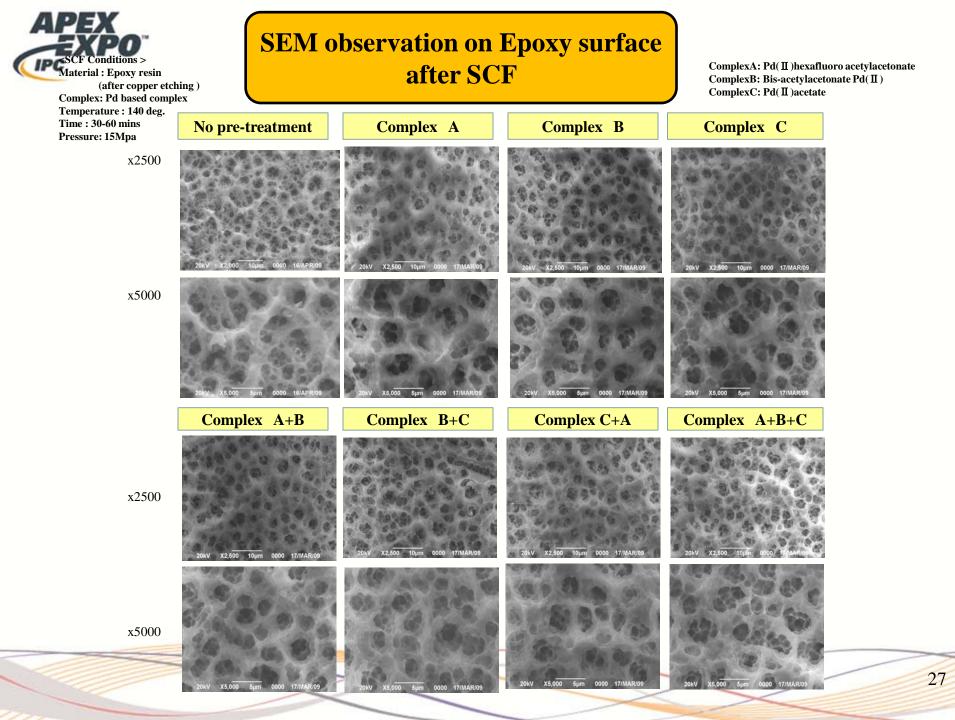


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## **Test details**

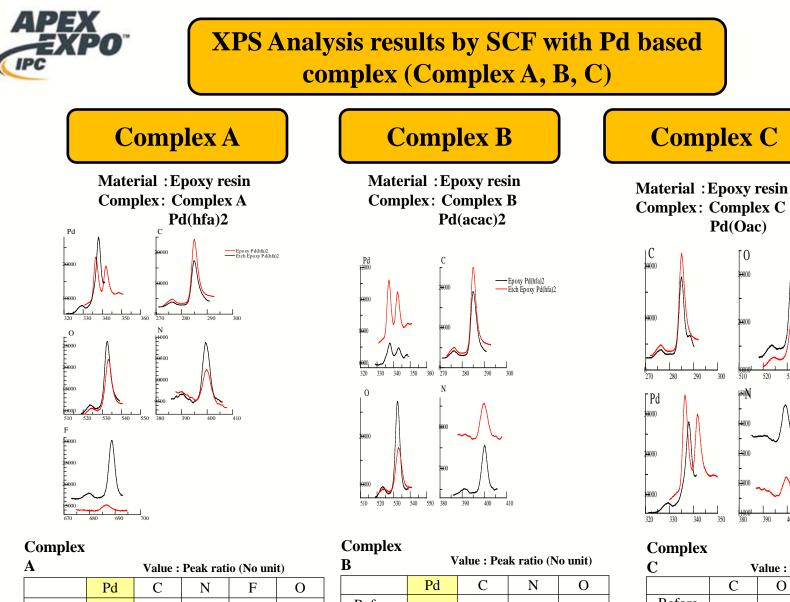
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C .	Value : Peak ratio (No unit)							
	С	C O Pd N						
Before Etching	1.00	0.72	0.17	0.48				
After Etching	1.00	0.46	0.17	0.32				

[0]

Epoxy Pd(OAc) Etch-Epoxy Pd(OAc)

Α	Value : Peak ratio (No unit)					
	Pd	C	N	F	0	
Before Etching	0.09	1.00	0.36	0.46	0.56	
After Etching	0.10	1.00	0.25	0.16	0.34	

Complex B	Value : Peak ratio (No unit)					
	Pd	С	N	0		
Before Etching	0.04	1.00	0.23	0.53		
After Etching	0.05	1.00	0.20	0.26		



## Peak ratio of Pd after SCF treatment with Pd based complexes

#### Complex

#### Value : Peak ratio (No unit)

Л					
	Pd	С	N	F	0
Etch前	0.09	1.00	0.36	0.46	0.56
Etch <b>後</b>	0.10	1.00	0.25	0.16	0.34
~ .					

#### Complex

B				
	Pd	С	Ν	0
Etch前	0.04	1.00	0.23	0.53
Etch後	0.05	1.00	0.20	0.26

#### Complex

1	7
Ľ	~
_	

C				
	C	0	Pd	Ν
Etch前	1.00	0.72	0.17	0.48
Etch <b>後</b>	1.00	0.46	0.17	0.32

#### Complex

A	+	·B	

	С	0	Pd	F
Etch前	1.00	0.62	0.08	0.35
Etch <b>後</b>	1.00	0.48	0.11	0.25

#### ComplexB+C

	С	0	Pd
Etch前	1.00	0.76	0.18
Etch後	1.00	1.15	0.51

#### ComplexC+A

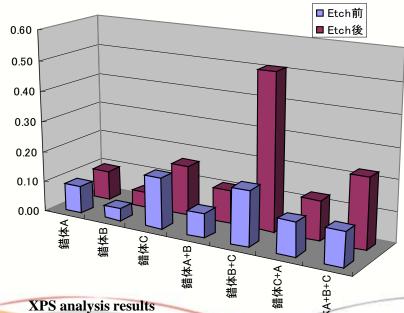
	C	0	Pd	F
Etch前	1.00	0.64	0.11	0.49
Etch後	1.00	0.50	0.13	0.24

#### ComplexA+B+

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J	Ļ	`	

	С	0	Pd	F
Etch前	1.00	0.59	0.11	0.32
Etch後	1.00	0.73	0.23	0.31

	Pd	Concentration	
Complex	Before Etch	After Etch	Fluctuation (%)
ComplexA	0.09	0.10	9
ComplexB	0.04	0.05	18
ComplexC	0.17	0.17	-2
ComplexA+B	0.08	0.11	38
ComplexB+C	0.18	0.51	183
ComplexC+A	0.11	0.13	18
ComplexA+B+C	0.11	0.23	109
Average	0.11	0.18	66



(Before Etching : blue and after Etching : purple ) (Pd concentration in case of assuming Carbon concentration 1.0 )



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- (4) Measurement of complex impregnation by XPS analysis

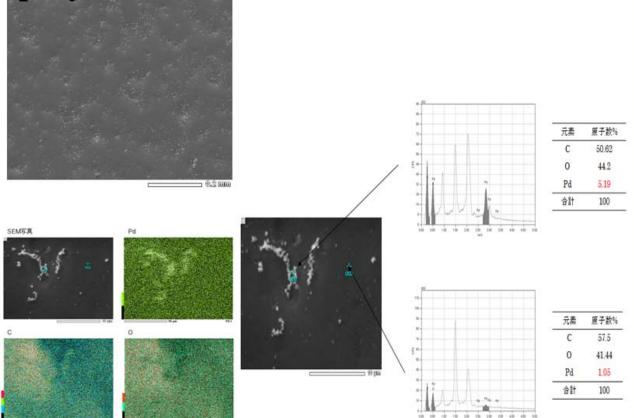
## (5)Epoxy EDX Mapping check

- (6) TEM Analysis for Epoxy Matte & Shiny Surface
- (7) Verification peel strength performance



### **Epoxy EDX Mapping**

#### **Epoxy Surface**



#### EDX Mapping x5,000



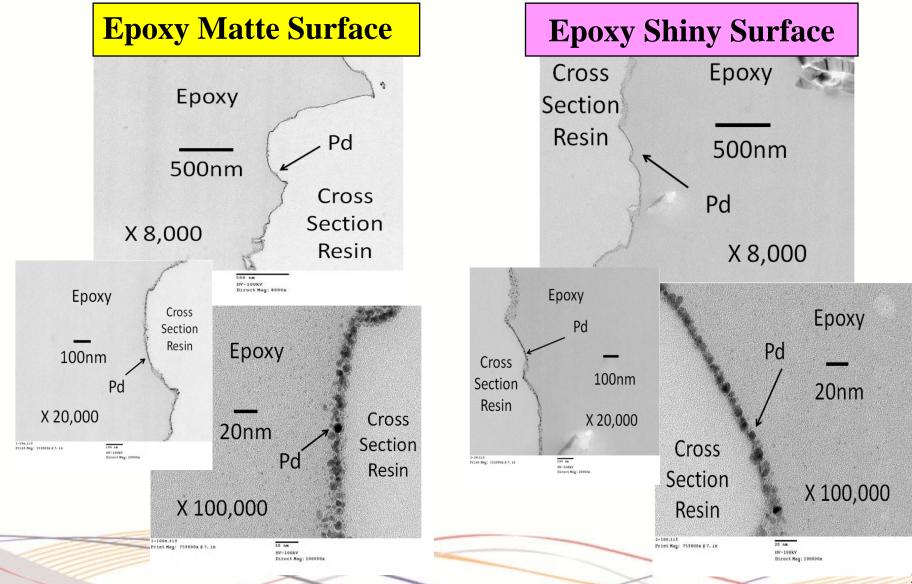
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## (6) TEM Analysis for Epoxy Matte & Shiny Surface

(7) Verification peel strength performance

Skin Impregnations by TEM Analysis for Epoxy



APEX

IPC

**XPO** 



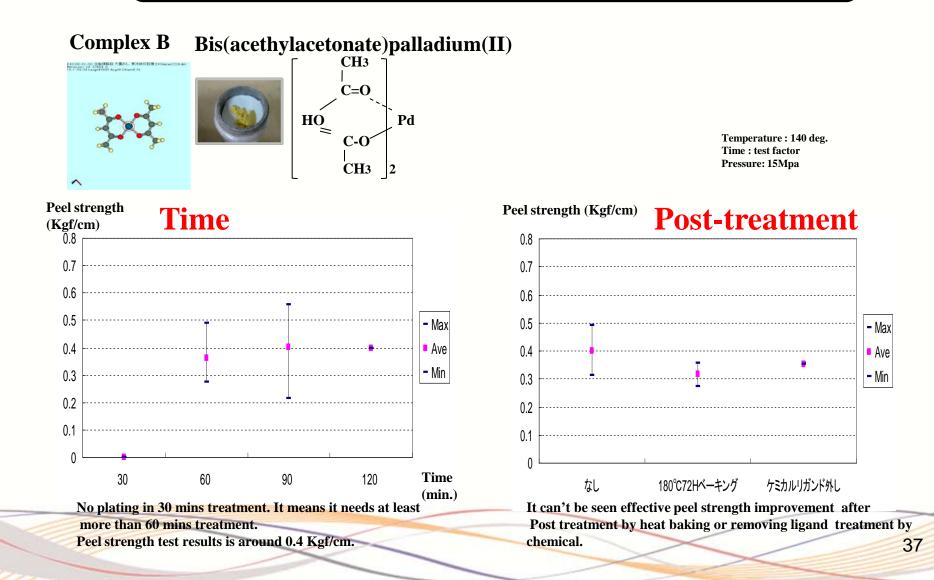
## **Test details**

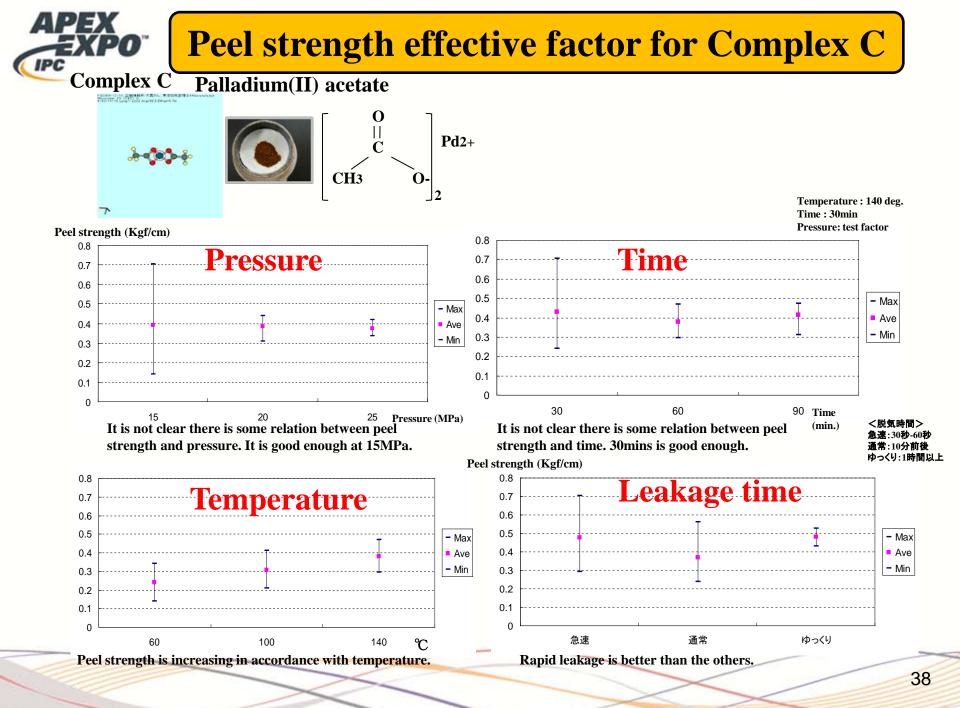
- (1) Application SCF technology to Epoxy resin
- (2) SCF treatment condition and Complex selection
- (3) SEM observation on Epoxy surface after SCF
- (4) Measurement of complex impregnation by XPS analysis
- (5) Epoxy EDX Mapping check
- (6) TEM Analysis for Epoxy Matte & Shiny Surface

## (7) Verification peel strength performance



## **Peel strength effective factor for Complex B**





## **Complex properties from test results**

Consideration the effectiveness to peel strength by complex type.

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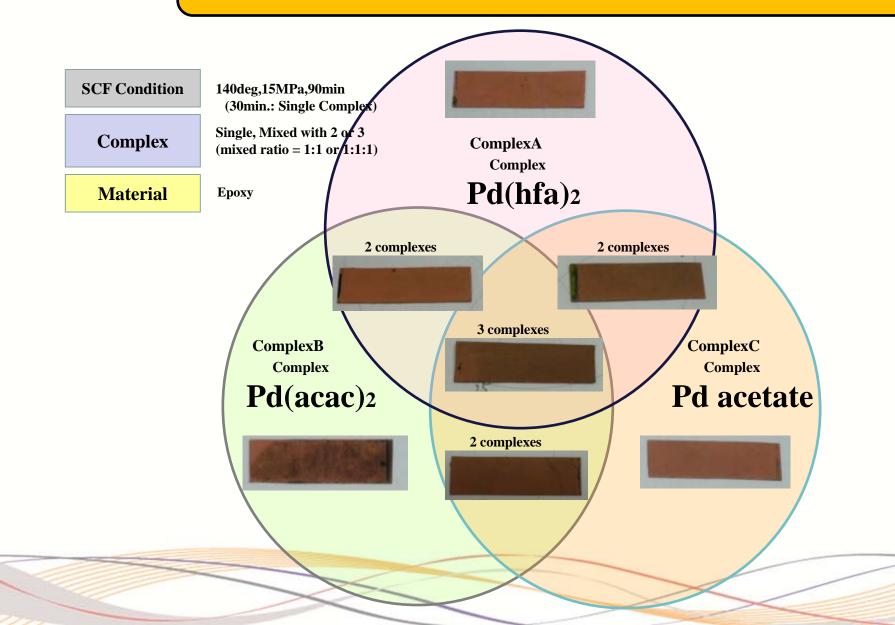
EXPO

	Complex	Pressure	Temp.	Time	Leakga ge	
	Pd(II)hexafluoro acetylacetonate       Dark Blown     Complex A       powder	0		×		O : Effective × : Not effective ∆ : Unknown — : Unconfirmed
Pd 系	Bis-acetylacetonate Pd(II) Yellow powder Complex B			0		
	Pd(II)acetate Red blown powder Complex C	×	0	×	Δ	

	Complex	Complex feature	After SCF	Dissolution to CO2
	Pd(II)hexafluoro acetylacetonate       Dark blown       powder   Complex A	Dependence on pressure. High pressure tends to be higher peel strength		High. Most of all were consumed after SCF
Pd 系	Bis-acetylacetonate Pd(II) Yellow Complex B powder	Short time treatment is not good enough for plating. It requires at least 1 hour treatment for plating		Low. Most of them are still remaining after SCF. It changed coloring
W	Pd(II)acetate Dark blown powder C	As a whole, peel strength is higher level and performance is good. Better to be combine to another complex		Low. Hardly consumed after SCF.



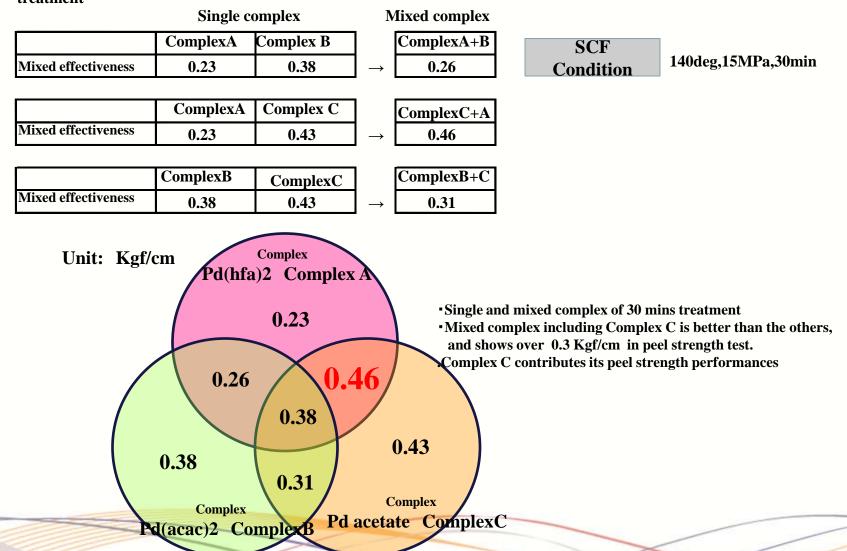
## **Appearance after SCF & Electroless plating**





## **Peel strength test result**

Consideration of the differences single complex and mixed complex treatment

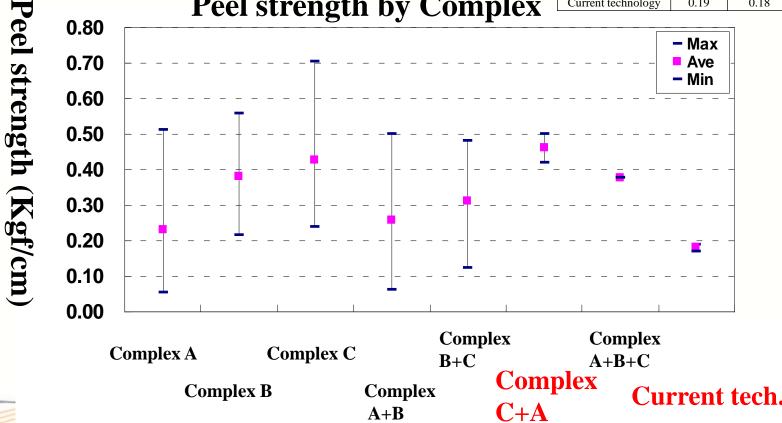




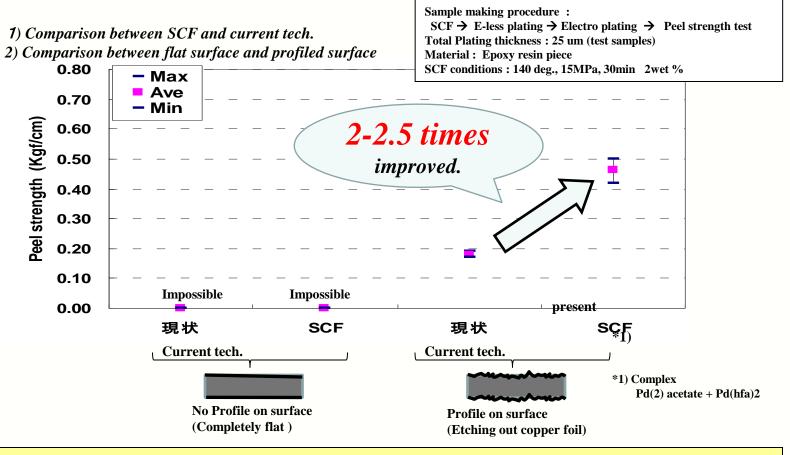
## **Comparison of peel strength test result**

	Max	Ave	Min
Complex A	0.51	0.23	0.06
Complex B	0.56	0.38	0.21
Complex C	0.70	0.43	0.24
Complex A+B	0.50	0.26	0.06
Complex B+C	0.48	0.31	0.12
Complex C+A	0.50	0.46	0.42
Complex A+B+C	0.38	0.38	0.38
Current technology	0.19	0.18	0.17

**Peel strength by Complex** 



## **Peel strength to Epoxy resin by SCF**



- 1) In applying SCF technology, Peel strength can be expected 2.0 -2.5 times stronger than existing process technology.(on epoxy resin board without copper)
- 2) However, it requires surface roughness on the epoxy resin. (not good enough for keeping peel strength on flat surface)

IPC



## **PCB Sample making status by SCF**

#### First time: L/S design= 50/50µm ⇒ Actual L/S= 50/50µm

# Image: line in the section of the s

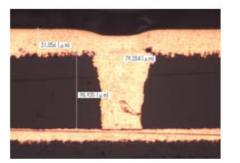
SCF technology

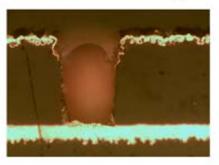
#### Current technology



**Circuit Conductors** 

L/S=50/50 µm





Insulation thickness:100µm, Laser VIA diameter : 50µm, Aspect ratio : 2.0



## **Conclusions and Future Examine Items**

#### As a effective results

- **1:It has recognized that SCF technology works effectively as a method of impregnation to epoxy resin.**
- 2:Copper peel strength on Epoxy resin by complex impregnation of SCF can make it possible to improve 2.0 -2.5 times better than existing technology.

#### <u>As a issue</u>

- **3:**Peel strength performances will be affected by surface condition (issue of surface treatment on material )
- 4:Relation between complex impregnation volume to epoxy resin and Pd ion contribution to plating (affection by removing Ligand ) is not yet confirmed.
- 5:It would be important point that confirming to impregnate complex and ligand status condition in improving peel strength in future study.



## **The Nobel Prize in Chemistry 2010**

*''for palladium-catalyzed cross couplings in organic synthesis''* Richard F. Heck, Ei-ichi Negishi, Akira Suzuki

#### Akira Suzuki



Photo: Hokkaido University, Japan

#### Ei-ichi Negishi



Photo: Purdue University, USA

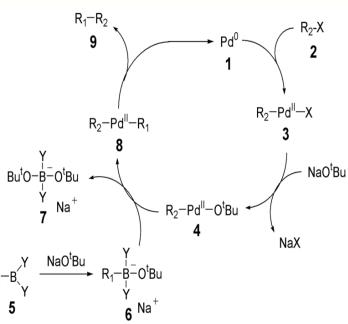
Richar<u>d F. Heck</u>



Photo: University of Delaware, USA

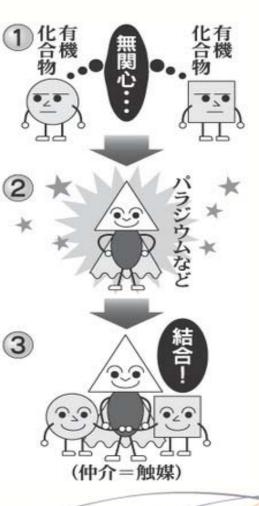
#### Suzuki/Miyaura Coupling Comprex Recycle

鈴木・宮浦カップリングの触媒サイクル



Negishi Coupling 根岸カップリング

 $R-ZnX + R'-Y \rightarrow R-R'$ 



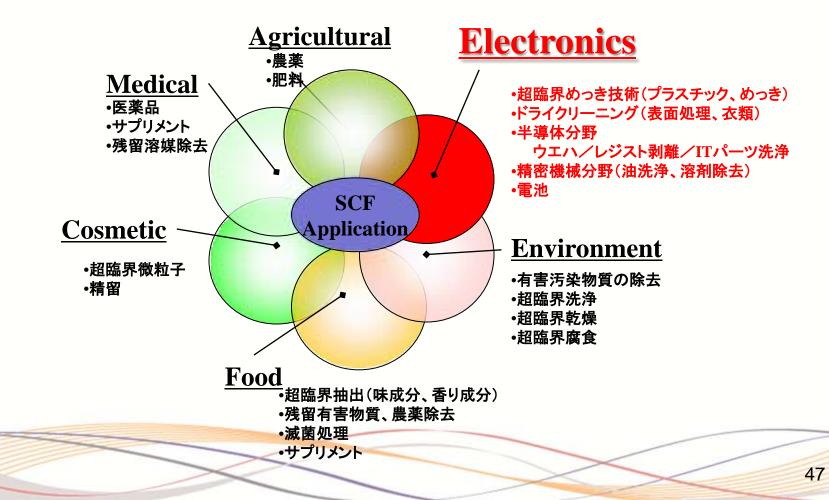


## **Application in various field**

a) Application: antenna, condenser, battery, Environmental technology 応用技術:アンテナ、コンデンサー、電池技術、環境技術ほか

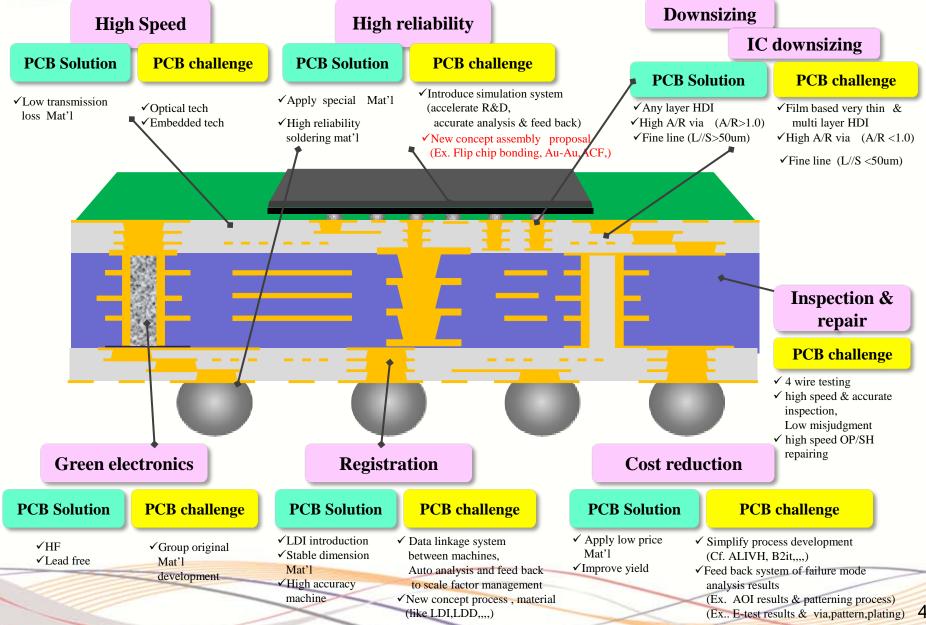
b) Application in various business field

各種ビジネス領域での応用



## **PCB solution– Challenge for core technology**

APEX





## Thank you

## Acknowledgements

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Meadville Enterprises (HK) Limited

美維企業(香港)有限公司

a member of TTM Technologies, Inc.